



# Certificate of Analysis

Standard Reference Material<sup>®</sup> 3184

Bromide Anion Standard Solution

Lot No. 020701

This Standard Reference Material (SRM) is intended primarily for use in anion ion chromatography or any other analytical technique that requires aqueous standard solutions for calibration or as control samples. One unit of SRM 3184 consists of five 10 mL sealed borosilicate glass ampoules of a single component solution prepared gravimetrically to contain a nominal 1000 mg/kg of bromide dissolved in filtered (0.22  $\mu$ m) 18 M $\Omega$  water.

Certified Value of Bromide: 1000 mg/kg  $\pm$  2 mg/kg at 22  $^{\circ}$ C

The certified value is based on: (1) gravimetric preparation using high purity potassium bromide, and (2) ion chromatography (IC).

The uncertainty in the certified value is calculated as

$$U = ku_c \text{ (mg/kg)}$$

where  $k = 2.32$  is the coverage factor for a 95 % confidence interval and  $u_c$  is the combined standard uncertainty. The combined standard uncertainty,  $u_c$ , has been calculated according to the ISO Guide [1]. The uncertainty in the certified value includes a source for the uncertainties for each method, gravimetric and chromatographic, and a source that accounts for possible biases among the methods. The uncertainty calculation procedure is described in Reference [2].

**Expiration of Certification:** The certification of **SRM 3184 Lot No. 020701** is valid, within the measurement uncertainty specified, until **29 August 2007**, provided the SRM is handled in accordance with instructions given in this certificate (see *Instructions for Use*). This certification is nullified if the SRM is damaged, contaminated, or modified.

**Maintenance of Certification:** NIST will monitor representative solutions from this SRM lot over the period of its certification. If substantive changes occur that affect the certification before the expiration of certification, NIST will notify the purchaser. Return of the attached registration card will facilitate notification.

This SRM was prepared gravimetrically and analyzed using ion chromatography by J.M. Smeller of the NIST Analytical Chemistry Division.

Statistical consultation was provided by W.F. Guthrie of the NIST Statistical Engineering Division.

The support aspects involved in the preparation, certification, and issuance of this SRM were coordinated through the NIST Standard Reference Materials Group by C.S. Davis.

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Certificate Issue Date: 07 January 2003

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## INSTRUCTIONS FOR USE

**CAUTION:** This SRM is an aqueous solution contained in tip-sealed borosilicate glass ampoules with prescored stems. Therefore, all appropriate safety precautions, including use of gloves during handling, should be taken to avoid accidental breakage or spillage. Unopened ampoules should be stored under normal laboratory conditions in an upright position inside the original container supplied by NIST.

**Opening an Ampoule:** When an ampoule is to be opened, that area of the stem where the prescored band is located (~5 mm below the encircling metallic band) should be carefully wiped with a clean, damp cloth and the body of the ampoule wrapped in absorbent material. Then holding the ampoule steady and with thumb and forefinger grasping the stem at the metallic band, **minimal** thumb pressure should be applied to the stem to snap it. Correctly done, the stem should break easily where prescored. Use of a metal file to break the stem is **NOT** recommended.

**Working Standard Solution:** After opening the ampoule, the entire contents must be transferred immediately to an empty, dry, weighed polyethylene bottle before *working standard solutions* are prepared. If an acid washed bottle is used, the bottle must be rinsed three times, soaked for at least 24 h in 18 M $\Omega$  water, and then thoroughly dried prior to use. Working standard solutions in the range of 10 mg/kg to 100 mg/kg, from which more dilute standards can be prepared, are recommended. The user should establish internal laboratory procedures that specify a maximum shelf life for a working standard solution. Two procedures for the preparation of working standard solutions follow.

**Preparation of Working Standard Solutions by Mass:** Each working standard solution should be prepared by employing one or more ampoules of the SRM into an empty, dry, preweighed polyethylene bottle, and then reweighing the bottle. Filtered water must be added by mass to bring the solution to the desired dilution. The dilution need not be exact since the mass of the empty bottle, mass of the bottle plus SRM aliquot, and the final diluted mass of the solution will permit calculation of the exact concentration of the working standard solution. Dilutions prepared gravimetrically as described will need no correction for temperature and no further correction for true concentration in vacuum. The working standard solution concentration will be in mass fraction units (mass of bromide per mass of solution). Dilute SRM solution concentrations will be in mg/kg units. Volumetric dilutions are **NOT** recommended due to uncertainties in volume calibrations and variations in density. However, for user convenience, a procedure for volumetric preparation that will minimize the major sources of error is given below.

**Preparation of Working Standard Solutions by Volume:** Each working standard solution should be prepared by employing one or more ampoules of the SRM into an empty, dry, preweighed polyethylene bottle, and then reweighing the bottle. The solution must now be transferred to a Class A volumetric flask and the polyethylene bottle reweighed to determine the exact mass of SRM solution transferred. The solution in the flask is then diluted to 99 % + volume filtered using water, mixed thoroughly, and the remaining few drops needed to dilute to exact volume carefully added. The concentration (in mg/mL) of the resulting diluted working standard solution can then be calculated by multiplying the mass (in kg) of the SRM solution amount by the SRM certified value (in mg/kg) and dividing the numerical product by the calibrated volume (in mL) of the flask used for dilution. If this procedure is followed, no correction for density is needed, and although the concentration of the resulting working standard solution may be an uneven fraction of the original SRM concentration, it will be known as accurately as a volumetric dilution permits.

## REFERENCES

- [1] *Guide to the Expression of Uncertainty in Measurement*, ISBN 92-67-10188-9, 1st Ed., ISO, Geneva, Switzerland, (1993); see also Taylor, B.N. and Kuyatt, C.E.; *Guidelines for Evaluating and Expressing the Uncertainty of NIST Measurement Results*; NIST Technical Note 1297, U.S. Government Printing Office, Washington, DC (1994); (available at <http://physics.nist.gov/Pubs/>).
- [2] Levenson, M.S.; et. al; *An ISO GUM Approach to Combining Results From Multiple Methods*; J.Res. Nat'l. Inst. Stand. Tech., Vol. 105, No.4, pp. 571-579 (July-August 2000).

*Users of this SRM should assure that the certificate in their possession is current. This can be accomplished by contacting the SRM Group at: telephone (301) 975-6776; Fax: (301) 948-3730; e-mail: [srminfo@nist.gov](mailto:srminfo@nist.gov), or internet <http://ts.nist.gov/srm>.*